

**Subject card**

<b>Subject name and code</b>	Monographic lecture - Intermolecular interactions in bioinorganic systems, PG_00080896						
<b>Field of study</b>	Chemical Business						
<b>Date of commencement of studies</b>	February 2025	<b>Academic year of realisation of subject</b>			2025/2026		
<b>Education level</b>	Master's studies	<b>Subject group</b>			Obligatory subject group in the field of study Optional subject group		
<b>Mode of study</b>	full-time studies	<b>Mode of delivery</b>			at the university		
<b>Year of study</b>	2	<b>Language of instruction</b>			Polish		
<b>Semester of study</b>	3	<b>ECTS credits</b>			3.0		
<b>Learning profile</b>	academic	<b>Assessment form</b>			credit		
<b>Conducting unit</b>	Department of Bioinorganic Chemistry -> Faculty of Chemistry -> Rector						
<b>Name and surname of lecturer (lecturers)</b>	<b>Subject supervisor</b>		dr hab. Aleksandra Dąbrowska				
	<b>Teachers</b>						
<b>Lesson types</b>	<b>Lesson type</b>	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	<b>Number of study hours</b>	30.0	0.0	0.0	0.0	0.0	30
	E-learning hours included: 0.0						
<b>Learning activity and number of study hours</b>	<b>Learning activity</b>	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	<b>Number of study hours</b>	30		5.0		40.0	75
<b>Subject objectives</b>	<ol style="list-style-type: none"> <li>1. Understanding of basic concepts and bonds/interaction concepts in bioinorganic chemistry.</li> <li>2. Indication of the role of bonds/interactions and their influence on the properties of matter and selected biochemical processes.</li> <li>3. Basics of Pharmacokinetics and LADME profile</li> </ol>						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[BCHMU2_W05] Knows and understands the main trends in the development of chemistry combined with economics as two interpenetrating scientific disciplines.	Knows the basic assumptions of pharmacokinetics and methods for designing drugs based on knowledge of biochemical mechanisms.	[SW4] test/exam - oral or written
	[BCHMU2_U02] Is able to define her/his interests, develop them within the chosen direction and in connection with the subject of her/his master's thesis by implementing the process of self-education and planning her/his professional career.	Has the ability to obtain information from various source materials, including information obtained from electronic databases.	[SU4] test/exam - oral or written
	[BCHMU2_U01] Is able to, on the basis of her/his knowledge, propose a solution to problems in chemistry, taking into account the economic aspect by using advanced measurement techniques.	Is able to determine basic physicochemical properties and reactivity chemical compounds based on their composition and structure.	[SU4] test/exam - oral or written
	[BCHMU2_W01] Knows and understands complex physicochemical processes and is able to analyse their course in connection with other fields of science.	Has general knowledge of basic concepts, principles and theories of bonds and interactions also in relation to general natural phenomena.	[SW4] test/exam - oral or written
[BCHMU2_K04] Is willing to properly assess the acquired knowledge, respect and disseminate it in order to solve specific cognitive and practical issues.	Understands the importance of basic knowledge in the field of chemistry and the social aspects of practical application of acquired knowledge and skills, as well as the related responsibility.	[SK4] test/exam - oral or written	
Subject contents	Atoms and molecules and their surroundings. Theories of chemical bonds. Types of intermolecular interactions and their energy. Factors influencing the strength of bonds and interactions as well as physicochemical properties. Elements of pharmacokinetics.		
Prerequisites and co-requisites			
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	exam	51.0%	40.0%
	test-I	51.0%	30.0%
	test-II	51.0%	30.0%
Recommended reading	Basic literature	<ol style="list-style-type: none"> <li>1. P.A. Cox, <i>Krótkie wykłady, chemia nieorganiczna</i>, PWN Warszawa (2003).</li> <li>2. F.A. Cotton, G. Wilkinson, P.L. Gaus, <i>Chemia nieorganiczna - podstawy</i>, PWN Warszawa (1995).</li> </ol>	
	Supplementary literature	<ol style="list-style-type: none"> <li>1. N.N. Greenwood, A. Earnshaw, <i>Chemistry of the elements</i>, Pergamon, wyd. II (2005).</li> <li>2. C.E. Housecroft, A.G. Sharpe, <i>Inorganic chemistry</i>, Pearson, Prentice Hall, Ed III (2008).</li> <li>3. S.J. Lippard, J.M. Berg, <i>Podstawy chemii bionieorganicznej</i>, PWN Warszawa (1998).</li> <li>4. I.G. Kaplan, <i>Intermolecular Interactions</i>, chap. 1,2,5, Wiley (2006).</li> <li>5. P. Schuster, G. Zundel and C. Sandorfy, Eds., <i>The Hydrogen Bond, Recent Developments in Theory and Experiments</i>, North Holland (1976).</li> <li>6. Publikacje tematyczne wskazane przez prowadzącego.</li> </ol>	
	eResources addresses	Basic <a href="https://bg.ug.edu.pl/">https://bg.ug.edu.pl/</a> - books and e-books Supplementary <a href="https://bg.ug.edu.pl/">https://bg.ug.edu.pl/</a> - books and e-books	
Example issues/ example questions/ tasks being completed			
Work placement	Not applicable		

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