

Subject card

Subject name and code	Inorganic chemistry, PG_00080831						
Field of study	Chemical Business						
Date of commencement of studies	October 2025		Academic year of realisation of subject		2025/2026		
Education level	Bachelor's studies		Subject group		Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	full-time studies		Mode of delivery		at the university		
Year of study	1		Language of instruction		Polish		
Semester of study	2		ECTS credits		2.0		
Learning profile	academic		Assessment form		credit		
Conducting unit	Laboratory of Physicochemistry of Coordination Complexes -> Department of General and Inorganic Chemistry -> Faculty of Chemistry -> Rector						
Name and surname of lecturer (lecturers)	Subject supervisor		dr hab. Dariusz Wyrzykowski				
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	0.0	0.0	30.0	0.0	0.0	30
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	30		5.0		15.0	50
Subject objectives	consolidation of basic theoretical knowledge in the field of inorganic chemistry introduction to important problems of contemporary inorganic chemistry presentation of the most important contemporary issues in inorganic chemistry that constitute progress in this field developing the ability to independently experiment and interpret the obtained results and solve problems when conducting chemical experiments						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[BCHINŻ_K04] Demonstrates responsibility for the safety of her/his own and others' work.	Demonstrates responsibility for the safety of one's own and others' work.	[SK1] oral statement/conversation/discussion
	[BCHINŻ_U08] Uses the chemical nomenclature and engineering terminology properly.	Appropriately uses chemical nomenclature and engineering terminology.	[SU3] text preparation/written work
	[BCHINŻ_U03] Plans, selects the appropriate research and measuring equipment and performs simple chemical experiments; analyses the results and draws conclusions based on them.	Plans, selects appropriate research and measurement equipment and apparatus, and performs chemical experiments; analyzes the results and formulates conclusions based on them.	[SU5] implementation of a problem task
	[BCHINŻ_U09] Using the acquired knowledge, skills and various sources of scientific information independently prepares written papers and oral presentations.	Using the acquired knowledge, skills and various sources of scientific information, he independently prepares written works and oral presentations.	[SU3] text preparation/written work
	[BCHINŻ_U02] Uses basic methods, techniques and tools in formulating and solving engineering tasks in the field of chemistry.	Applies methods, techniques and tools in formulating and solving engineering tasks in the field of chemistry.	[SU4] test/exam - oral or written
	[BCHINŻ_W10] Applies safety and hygiene principles when working on a test and measurement stand or in the field.	Knows and understands the principles of safety and hygiene when working at a research and measurement station or in the field.	[SW1] oral statement/conversation/discussion
	[BCHINŻ_W02] Enumerates basic laws and theories in chemistry, physics and mathematics necessary to formulate and solve simple engineering tasks.	Enumerates basic laws and theories in chemistry, physics and mathematics necessary to formulate and solve simple engineering tasks.	[SW3] text preparation/written work
Subject contents	Topics of the lecture: periodicity and the chemistry of the elements, physicochemical properties of inorganic and coordination compounds. The following items are included: periodicity, chemical bonding, coordination compounds, types of chemical reactions, properties of chemical elements and their compounds. The groups of elements are presented in the following order: group 1, group 2, group 13, group14, group 15, group 16, group 17, group 18, and d-elements (groups 3-12; first transition row, second transition row, and third transition row). Topics of auditory classes: basic types of inorganic compounds, valence bond theory, hybridization and molecular geometry; molecular orbital theory; solid state bonds: ionic, covalent, metallic; metals, semiconductors and insulator; coordination compounds. Topics of lab classes: investigation of physicochemical properties of the elements, inorganic and coordination compounds based on chemical experiments.		
Prerequisites and co-requisites			
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
		51.0%	100.0%
Recommended reading	Basic literature	A. Bielański Podstawy chemii nieorganicznej, PWN 2002 J. D. Lee Związła chemia nieorganiczna, PWN 1997 L. Jones, P. Atkins Chemia ogólna, PWN 2004 B. Literatura uzupełniająca	
	Supplementary literature	L. Pajdowski Chemia ogólna, PWN 1999	
	eResources addresses		
Example issues/ example questions/ tasks being completed	Explain the differences in the magnetic properties (diamagnetic, paramagnetic) of sodium peroxide and potassium superoxide (tip: based on the analysis of the distribution of electrons in the molecular orbitals of the peroxygen ion and the superoxide ion, indicate which of the compounds has unpaired electrons/electron. On this basis, indicate which of the compounds has paramagnetic properties and which has diamagnetic properties).		
Work placement	Not applicable		

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